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## Regio and Enantioselective Synthesis of 4-Carbomethoxymethyl-1,4 Dihydropyridines

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Abstract: Regio and enantioselective 1,4 addition of ethoxyvinyl copper reagent on a pyridine bearing in position 3 a chiral aminal is described. This reaction allows the synthesis of various chiral N-acyi-4-methoxymethyl-1,4-dihydropyridines.

Regio and enantioselective introduction of carbalkoxymethyl group onto a functionalized pyridine is an important challenge for a synthetic point of view. A recent paper dealing with such subject in achiral version<sup>1</sup> prompted us to present our results on this field. From our part, we were interested by synthesis of chiral 4-carbalkoxymethyl-1,4-dihydropyridines 1. Recently, we have proposed a synthesis of chiral 4substituted-1,4-dihydropyridines 2 (C4 adduct) from chiral aminal 3 by a regio and diastereoselective 1,4addition of organocopper reagents in the presence of an acylating reagent <sup>2</sup> (Scheme 1 and entry 1 in the Table). Among the several functionalized organocopper used, it was found that the reagent CuCH<sub>2</sub>CO<sub>2</sub>Et afforded exclusively the C6 adduct 4 (Scheme 1) whatever the acylating reagent (methyl chloroformate or pivaloyl chloride, entry 2).



Some 1,4 selectivity has been reported by addition of the Reformastky reagent derived from ethyl bromoacetate, on N-acyl pyridinium salts.<sup>3</sup> In our case, in the presence of methyl chloroformate, this reagent was found to be unreactive (entry 3).

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Entry	R <sup>1 a)</sup>	R	C4 / C6	Yield (%) <sup>b)</sup>
1	R <sup>1</sup> Cu (R <sup>1</sup> = primary or secondary alkyl, vinyl, aryl (ref 2)	CH <sub>3</sub> O, CH <sub>3</sub>	100 / 0	77-95
2	CuCH <sub>2</sub> CO <sub>2</sub> Et (ref 2)	CH <sub>3</sub> O, tBu	0 /100	20-25
3	BrZnCH <sub>2</sub> CO <sub>2</sub> Et	CH₃O	-	-
4		CH₃O	0 /100	80
5	LiCH <sub>2</sub> CS <sub>2</sub> Me	CH <sub>3</sub> O	-	-
6	LiCu(CH <sub>2</sub> CS <sub>2</sub> Me) <sub>2</sub>	CH₃O	0 /100	30
7	CuCH(SiMe <sub>3</sub> )CO <sub>2</sub> El <sup>c)</sup>	CH₃	23 / 77	50 <sup>d)</sup>
8		´ Ph	10/90	70 <sup>d)</sup>
9		tBu	-	-
10		CH₃O	15 / 85	80
11		Ph	90 / 10	75

a) All the reactions were performed in THF from - 70°C to 0°C except for entry 4. In this case, the reaction was performed in  $CH_2CI_2$  at R.T. b) For all the C6 adducts, de = 0. c) Prepared with an excess of soluble copper salt (2 CuBr,Me<sub>2</sub>S, 4 LiBr). d) The de of the C4 adduct was not determined.

By analogy to the observed regioselective 1,4 addition of silyl enol ethers onto N-acyl pyridinium salts<sup>4</sup>, a silyl ketene acetal<sup>5</sup> was opposed to the aminal 3 in the presence of methyl chloroformate in CH<sub>2</sub>Cl<sub>2</sub> to give exclusively the C6 adduct<sup>6</sup> (entry 4).

The soft reagent  $Cu(CH_2CS_2Me)_2^7$ , obtained from the corresponding lithio derivative<sup>8</sup> (which gave the starting marerial, entry 5), yielded also, in the presence of methyl chloroformate, exclusively to the C6 adduct (entry 6).

1,4-additions of lithium salt of silyl acetates on enones<sup>9</sup> and N-alkylpyridinium salts,<sup>10</sup> has been reported. Therefore a conjugate addition was attempted with a silyl copper enolate (entry 7) on 3 in the presence of acetyl chloride. A mixture of C4 and C6 adducts was obtained in a 23/77 ratio (entry 7). Attempts to increase the C4 selectivity by changing the acyl chloride <sup>11</sup> were unsuccessfull (entries 8 and 9).

We decided then to use the ethoxyvinyl copper  $^{12}$  which is an equivalent of <sup>-</sup>CH<sub>2</sub>-CHO. With this reagent, in the presence of methyl chloroformate (entry 10), a mixture of C4 and C6 adducts was obtained in a 15/85 ratio. On the other hand with benzyl chloride the C4 adduct was obtained with a very good selectivity (90/10, entry 11). No explanations were found for this unexpected regioselectivity.

The diastereomeric purity of the so obtained 1,4-dihydropyridine was measured by <sup>1</sup>H NMR and found to be good (de = 80%). Furthermore, the two diastereomers are easily separable by flash chromatography (SiO<sub>2</sub>, C<sub>6</sub>H<sub>12</sub>/Et<sub>2</sub>O,80/20) to give the pure major diasteromer 5 in 67% yield (Scheme 2). A mild acidic hydrolysis (HCl 2%) afforded the monoaldehyde 6 and, by more forcing the conditions (HCl 5%), the dialdehyde 7<sup>13</sup> was obtained in good yield (Scheme 2). The optical purity of 7 was measured by <sup>1</sup>H NMR of the corresponding monoaminal 8 and found to be excellent (ee > 95%). Finally, the dialdehyde 7 was oxidized <sup>14</sup> to give the desired ester 9<sup>15</sup> (Scheme 2).





For synthetic purposes, it was interesting to test the possibility to exchange the N-acyl group on aminal 5 where the aldehyde function is still protected. Addition of 3 equivalents of BuLi on 5, in THF at -30°C, gave the corresponding amide which was directly acylated by a new acyl chloride to give, in a good yield, the dihydropyridine 10. This compound was hydrolyzed into the dialdehyde  $11^{16}$  and then, transformed into the corresponding aminal 12 in order to measure the optical purity. Unfortunately, this was not possible by <sup>1</sup>H NMR (presence of conformers), and 12 was converted into 13 by alcaline hydrolysis. An accurate measure was then possible and the de was found to be exactly the same than the starting aminal 5 (de > 95%). Therefore no epimerisation occurs during all the procedure.



Scheme 3

In conclusion it is possible to introduce selectively in 4 position an equivalent of - CH<sub>2</sub>CHO which is a precursor of the corresponding ester. Furthermore the possibility of changing the N-acyl group allows usefull synthetic applications.

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## **References** and notes

- 1 Murall Dhar, T.G.; Gluchowski, C., Tetrahedron Lett 1994, 35, 989-992
- Gosmini, R.; Mangeney, P.; Alexakis, A.; Commerçon, M., Normant, J.F., Synlett. 1991, 111-113. Mangeney, P.; Gosmini, R.; Alexakis, A., Tetrahedron Lett 1991, 32, 3981-3984.
   Mangeney, P.; Gosmini, R.; Raussou, S.; Commerçon, M.; Alexakis, A., J.Org. Chem. 1994, 1877-1888
- 3 Courtois, G.; Arnaout, A.L.; Miginiac, L., Tetrahedron Lett 1985, 26, 1027-1030
- Akiba, K.Y.; Nishihara, Y.; Wada, M., Tetrahedron Lett 1983, 24, 5269-5272. Wada, M.;
  Nishihara, Y.; Akiba, K.Y., Tetrahedron Lett 1985, 26, 3267-3270. Akiba, K.Y.; Ohtani, H.;
  Yamamoto, Y., J.Org. Chem. 1986, 51, 5328-5332
- Onaka, M.; Ohno, R.; Izumi, Y., Tetrahedron Lett 1989, 30, 747-750
- 5 Bellasoued, M.; Gaudemar, M., Tetrahedron Lett 1990, 31, 209-212
- 6 The resulted C6 adduct was directly transformed into the ethyl ester (EtOH, DCC, DMAP, 12 hr.)
- 7 The copper derivative was prepared by addition of 1/2 CuI on the lithium salt prepared according ref. 8
- 8 Metzner, P., Synthesis 1992, 1185-1194
- 9 Oppolzer, W.; Guo, M.; Baettig, K., Helv. Chem. Acta, 1983, 2140-2144
- 10 Wenkert, E.; Angell, E.C.; Drexler, J.; Moeller, P.D.R.; Pyrek, J. St.; Shi, Y.J.; Sultana, M.; Vankar, Y.D., J. Org. Chem., 1986, 51, 2995-3000
- 11 Comins, D.L.; Abdullah, A.H., J. Org. Chem., 1982, 47, 4315
- Obtained by addition of an excess of soluble copper salt (2CuBr, Me<sub>2</sub>S, 4LiBr) in THF, to the corresponding lithio derivative prepared according : Kreisler, S.Y.; Schlosser, M., J. Org. Chem., 1978, 43, 1595-1597
- 13  $[\alpha]_D^{25} = -151 \ (c = 1.5, CHCl_3); {}^{1}H \ NMR \ (CDCl_3, 200 \ MHz) \ \delta \ 9.6 \ (t, \ J = 1.5 \ Hz, 1H), 9.2 \ (s, 1H), 7.6 \ (s, 1H), 7.5 \ (z, 1H), 6.8 \ (d, \ J = 8 \ Hz, 1H), 5.2 \ (dd, \ J = 8 \ Hz, \ J = 5Hz, 1H), 3.7 \ (m, 1H), 2.8 \ and 2.5 \ (z \ ddd, \ J = 17.3, \ J = 8 \ Hz, \ J = 4.2 \ Hz, \ J = 1.5 \ Hz, 2H); {}^{13}C \ NMR \ (100 \ MHz) \ 200.3, \ 190.7, \ 167.6, \ 142.9, \ 132.3, \ 131.9, \ 128.8, \ 123.6, \ 121.8, \ 112.4, \ 49.5, \ 26.5.$
- 14 Bal, B.S.; Childers, W.E. jr.; Pinnick, H.W., Tetrahedron 1981, 37, 2091-2096
- 15  $[\alpha]_D^{25} = -97 \ (c = 6.5, \text{ CHCl}_3); ^1\text{H NMR (CDCl}_3, 200 \text{ MHz}) \delta 9.3 \ (s, 1\text{H}), 7.7 \ (s, 1\text{H}), 7.7-7.3 \ (m, 5\text{H}), 6.9 \ (d, J = 8.2 \text{ Hz}, 1\text{H}), 5.3 \ (dd, J = 8.2 \text{ Hz}, J = 4.8 \text{ Hz}, 1\text{H}), 3.6 \ (m, 1\text{H}), 3.3 \ (s, 3\text{H}), 1.98 \ \text{and} 1.75 \ (2 \ ddd, J = 13.8 \text{ Hz}, J = 8.4 \text{ Hz}, J = 6.2 \text{ Hz}, J = 5.4 \text{ Hz}, J = 4.2 \text{ Hz}, 2\text{H}); ^{13}\text{C}$ NMR (100 MHz) 190.7, 167.6, 141.9, 132.1, 128.9, 128.7, 123.9, 123, 113.6, 102, 53.3, 38.8, 27.7
- 16  $[\alpha]_D^{25} = -485 \ (c = 1.1, CHCl_3, ee = 80\%); {}^{1}H \ NMR \ (CDCl_3, 200 \ MHz) \ \delta \ 9.7 \ (t, \ J = 1.5 \ Hz, 1H), 9.4 \ (m, 1H), 7.9 \ (m, 1H), 6.8 \ (m, 4H), 5.3 \ (m, 1H), 3.8 \ (m, 9H), 2.8 \ and 2.5 \ (2 \ ddd, \ J = 17.3 \ Hz, \ J = 9.8 \ Hz, \ J = 5.3 \ Hz, \ J = 1.3 \ Hz, \ J = 0.5 \ Hz, 2H); {}^{1}SC \ NMR \ (100 \ MHz) \ 200.9, 167.5, 157.5, 149.4, 148.5, 142, 124.5, 123, 120.9, 112.9, 111.6, 111.4, 55.9, 49.5, 40.2, 29.6, 26.1$

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